March 1923 AMERICAN PHARMACEUTICAL ASSOCIATION

6. Since the previous work gave M. T. D. = 320 mg./Kg. it has been shown that the maximum decrease of toxicity due to amino acid, pure hydrosulphite and centrifugation is 40-60 mg./Kg.

Contribution from the Laboratories of the Upjohn Company, Kalamazoo, Michigan.

## MISCELLANEOUS CHEMICAL PAPERS.\* 4. Fractionation of Turpentine Oil.<sup>1</sup>

## BY W. F. SUDRO.

A quantity of approximately five gallons of turpentine was steam-distilled, the distillate being received in approximately liter fractions, sixteen in number. The distillates were immediately placed into bottles filled to the stopper in order to prevent oxidation or other changes. The data on the steam-distillation are given in Table I.

TABLE I.								
Steam distillation fraction.	Sp. gr. 20° C. (Westphal).	Refractive index 20° C.	Polarization 23° C.					
1	0.862	1.4660	+15.25					
2	0.863	1.4675	+14.85					
3	0.863	1.4671	+15.1					
4	0.864	1.4682	+14.7					
5	0.863	1.4680	+14.6					
6	0.863	1.4682	+14.5					
7	0.864	1.4661	+14.1					
8	0.863	1.4668	+13.9					
9	0.863	1.4670	+13.8					
10	0.864	1.4685	+13.3					
11	0.864	1.4684	+13.1					
12	0.864	1.4687	+12.4					
13	0.865	1.4690	+11.7					
14	0.865	1.4690	+10.5					
15	0.868	1.4698	+ 9.4					
16	0.870	1.4708	+7.4					

Upon inspection of the table it will be observed that the specific gravity gradually increases from the first fraction to the last, the constants running as one would expect. The polarization constants are also in strict agreement, the angle of rotation gradually diminishing with an increase in the number of the fractions, the only exception noted being either with fraction number two or three.

The indices of refraction show a relative increase with the exception of Fractions 7, 8 and 9. This will be discussed later.

\* From the Laboratory of Edward Kremers. Read before Scientific Section, A. Ph. A., Cleveland meeting, 1922.

<sup>1</sup> Part of thesis submitted for the degree of Master of Science at the University of Wisconsin, 1919.

The fractionation here recorded was carred out by Mr. Sudro for the purpose of gaining a better insight into the boiling temperature conditions of a so-called hydrocarbon oil. The results of a carefully performed series of steam and direct distillations of a commercial American turpentine oil may prove of interest to others, hence should be worth recording. The immediate incentive to perform such a "tedious" task was to obtain as pure a pinene fraction as obtainable by this means for chemical experiments that will be reported in another paper.—E. K.

22	20 JOURNAI, OF THE Vol. XII, No. 3																		
	16		÷	:		•	:	1.0	:	1.4688	1.0	:	1.4690	15.6	0.867	1.4698	82.4	0.873	1.4720
	15	distilled	:	÷	Nezligible amounts	÷	:	2.6	:	1.4680	9.9	0.863	1.4685	49.6	0.864	1.4687	37.9	0.868	1.4716
	14	Negligible amounts distilled	:	÷	Negligible	:	:	4.5	0.862	1.4676	15.2	0.863	1.4680	52.5	0.864	1.4683	27.8	0.868	1.4710
	13	Negligible	:	÷	1.0	:	1.4671	4.7	0.862	1,4674	24.5	0.863	1.4680	48.1	0.863	1.4682	21.7	0.871	1.4710
	12	1.0	:	1.4671	1.6	÷	1.4671	15.5	0.861	1.4677	35.6	0.863	1.4681	33.6	0.864	1.4686	12.8	0.874	1.4719
	11	2.1	•	1.4670	2.5	÷	1.4674	20.8	0.861	1.4676	36.9	0.862	1.4680	33.2	0.863	1.4689	4.5	0.878	1.4737
	10	1.5	:	1.4671	3.9	0.862	1.4673	21.1	0.862	1.4676	40.1	0.862	1.4680	30.4	0.864	1.4689	3.0	0.877	1.4731
	6	1.6	÷	1.4667	7.6	0.861	1.4672	20.8	0.862	1.4675	37.5	0.862	1.4679	25.2	0.864	1.4685	7.4	0.874	1.4723
II.	ж	2.6	:	1.4659	6.2	0.861	1.4669	27.7	0.861	1.4671	37.3	0.862	1.4677	19.0	0.864	1.4682	7.3	0.876	1.4721
T'able	٢	2.0	÷	1.4659	6.2	0.861	1.4663	37.6	0.861	1.4668	32.2	0.862	1.4677	16.9	0.864	1.4685	5.1	0.876	1.4728
	9	1.8	÷	1.4670	11.5	0.861	1.4671	35.5	0.861	1.4674	30.6	0.862	1.4679	16.6	0.864	1.4690	4.2	0.879	1.4742
	ъ	1.4	:	1.4663	8.8	0.861	1.4687	28.7	0.861	1.4673	35.8	0.862	1.4677	19.4	0.862	1.4689	6.0	0.873	1.4725
	4	1.2	:	1.4665	11.7	0.862	1.4675	25.7	0.863	1.4678	36.7	0.863	1.4680	20.6	0.864	1.4690	4.1	0.880	1.4750
	ñ	5.0	0.859	1,4646	21.7	0.859	1.4659	47.9	0.860	1.4670	14.8	0.861	1.4678	6.8	0.864	1.4687	3.9	:	1.4765
	64	લ લ	:	1 4648	28.9	0.861	1.4660	36.6	0.862	1.4670	17.0	0.863	1.4678	11.1	0.864	1.4687	4.4	÷	1.4740
	1	8.3	0.861	1.4654	31.0	0.861	1.4661	41.8	0.861	1.4668	0.6	0.862	1.4678	7.2	0.864	1.4687	2.9	:	1.4753
	Steam distillation fraction. Fraction boiling be- low 1550	Percentage of yield	Specific gravity at 20°	Fraction boiling at	Percentage of yield		20° Fraction boiling at	Percentage of yield	20°	Fraction boiling at 157 to 158°	Percentage of yield Specific gravity at	20°	20°. Fraction boiling at	Percentage of yield		Reiractive index at 20°	Percentage of yield	Specific gravity at 20°	Reifactive index at 20°

The next step was to distil each of the sixteen steam-distillation fractions over direct flame, fractions being obtained at the following boiling points: up to  $155^{\circ}$  C., 155 to  $156^{\circ}$ , 156 to  $157^{\circ}$ , 157 to  $158^{\circ}$ , 158 to  $160^{\circ}$ , and the residue distilling above  $160^{\circ}$ . The specific gravity and refractive index for each of the separate fractions were obtained. These, together with the percentage of yield, are tabulated in Table II.

The percentage of yield indicates the percent. by volume of the steam-distillation fractions which distilled at the given temperature. The results obtained are as expected in that the first steam-distillation fractions are richer in the lower than in the higher boiling-point fractions, while those steam-distillation fractions towards the last are richer in the higher than in the lower boiling-point fractions.

A mean is attained, however, with boiling-point fraction  $157-158^{\circ}$  where it will be noticed that the per cent. of that fraction gradually increases, then remains fairly constant, and finally decreases as the last steam-distillation fractions are reached.

The average specific gravity, the minimum and maximum values for the same, together with the difference between minimum and maximum, are included in Table III.

TABLE III.									
Steam distillation fractions.	Average specific gravity,	Minimum specific gravity.	Maximum specific gravity.	Difference between minimum and maximum values.					
1	0.862	0.861	0.864	0.003					
$^{2}$	0.863	0.861	0.864	0.003					
3	0.861	0.859	0.864	0.005					
4	0.866	0.862	0.880	0.018					
5	0.864	0.861	0.873	0.012					
6	0.865	0.861	0.879	0.018					
7	0.865	0.861	0.876	0.015					
8	0.865	0.861	0.876	0.015					
9	0.865	0.861	0.874	0.013					
10	0.865	0.862	0 877	0.015					
11	0.866	0.861	0.878	0.017					
12	0.865	0.861	0.874	0.013					
13	0.865	0.862	0.871	0.009					
14	0.864	0.862	0.868	0.006					
15	0.865	0.863	0.868	0.005					
16	0.870	0.867	0.873	0.006					

The same values with reference to the refractive index are given in Table IV.

With reference to Table IV, attention is called to the fact that the average values for the indices of refraction for the steam-distillation Fractions 7, 8 and 9, agree very closely with the constants for the neighboring fractions. This proves beyond a doubt that the variable values given in Table I for these fractions are to be accounted for by the fact that some disturbing factor must have been introduced. In other words, the abnormality was not due to the constituency of the turpentine itself.

Table V includes tabulated data with reference to the specific gravity and refractive index for each of the six boiling-point fractions. The values for the average specific gravity and refractive index are in strict accord with the results which would be expected.

	IABLE IV.									
Steam distillation fractions.	Average refractive index.	Minimum refractive index.	Maximum refractive index.	Difference between minimum and maximum values.						
1	1.4690	1.4654	1.4753	0.0099						
2	1.4681	1.4648.	1.4740	0.0092						
3	1.4684	1.4646	1.4765	0.0119						
4	1.4690	1.4665	1.4750	0.0085						
5	1.4687	1.4663	1.4725	0.0062						
6	1.4684	1.4670	1.4742	0.0072						
7	1.4680	1.4659	1.4728	0.0069						
8	1.4680	1.4659	1.4721	0.0062						
9	1.4684	1.4667	1.4723	0.0056						
10	1.4687	1.4671	1.4731	0.0060						
11	1.4688	1.4670	1.4737	0.0060						
12	1.4684	1.4671	1.4719	0.0048						
13	1.4683	1.4671	1.4710	0.0039						
14	1.4687	1.4676	1.4710	0.0039						
15	1.4692	1.4680	1.4716	0.0036						
16	1.4699	1.4688	1.4720	0.0032						
		TABLE V.								

•	1.	ABLE V.				
Boiling-point fractions,	155°.	1 <b>35 to 1</b> 56°.	156 to 157°.	157 to 158°.	158 to 160	°. 160°+.
Average sp. gr	0.860	0.861	0.861	0.862	0.864	0.874
Minimum sp. gr	0.859	0.859	0.860	0.861	0.862	0.871
Maximum sp. gr	0.861	0.862	0.863	0.863	0.867	0.880
Diff. betw. min. and max. values	0.002	0.003	0.003	0.002	0.005	0.009
Average refr. index	1.4662	1.4668	1.4674	1.4680	1.4687	1.4731
Minimum refr. index	1.4646	1.4659	1.4668	1.4677	1.4682	1.4731
Maximum refr. index	1.4671	1.4675	1.4688	1.4690	1.4698	1.4765
Diff. betw. min. and max. values	0.0025	0.0016	0.0020	0.0013	0.0016	0.0055
Sum total of distillates obt. for given						
temp	1.93%	8.94%	23.15%	25.99%	25.49%	14.53%

In Table V the sum total of the distillates obtained for the six boiling-point fractions is also given. Less than 10 per cent. of the entire volume of this particular sample of turpentine consisted of pinene, or the fraction boiling at 155 to  $156^{\circ}$ ; three-fourths distilling between 156 and  $160^{\circ}$ .

In conclusion, the data submitted show the relation between the several fractions, the results being in close agreement with the theoretical valuation. Slight discrepancies are noted but these are of minor consideration.

CONTRIBUTIONS TO THE KNOWLEDGE OF FOREST PRODUCTS.\*

IV. ON CYMENE FROM THE OIL OF MONARDA PUNCTATA L.

BY J. M. JOHNSTON, H. MERRITT, AND R. E. KREMERS.<sup>1</sup>

Introduction.—Among the plants whose volatile oil contains thymol is Monarda punctata L. It grows on poor sandy soil in nearly all states east of the Mississippi River. In the north at least it is often found associated with open stands of oak or jack pine. Hence this species of Labiatae deserves careful attention in connection

<sup>\*</sup> From the Laboratory for Organic Chemistry, Vanderbilt University.

<sup>&</sup>lt;sup>1</sup> At present Fellow in Biochemistry at the University of Wisconsin for the National Research Council. This incomplete paper is published as it is, because the work at the Vanderbilt Laboratory has been interrupted.